

# The effect of CO<sub>2</sub> on Earth's Radiation Budget

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March 28, 2012



M.I. Budyko (1969)\*

$$R \frac{\partial T}{\partial t} = Q_s(y)(1 - \alpha(y)) - \boxed{(A + BT)} - C \left( T - \int_0^1 T dy \right)$$

Outgoing Longwave Radiation OLR

- $A, B > 0$
- $T \nearrow \Rightarrow \text{OLR} \nearrow$
- $\text{CO}_2 \leftrightarrow A ?$



Ray Pierrehumbert

*“Big ideas come from small models.”*

May 23, 2011, Snowbird, UT

\*M.I. Budyko, The effect of solar radiation variations on the climate of the Earth, *Tellus* **21** (1969)

M.I. Budyko (1969)

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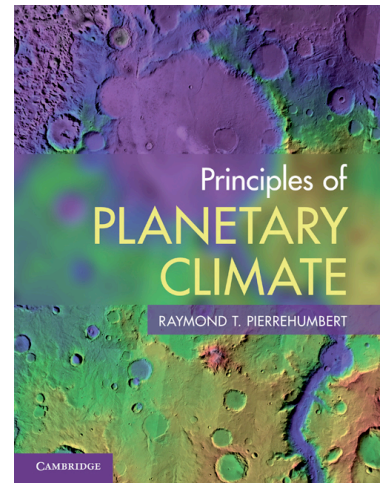
Outgoing Longwave Radiation OLR

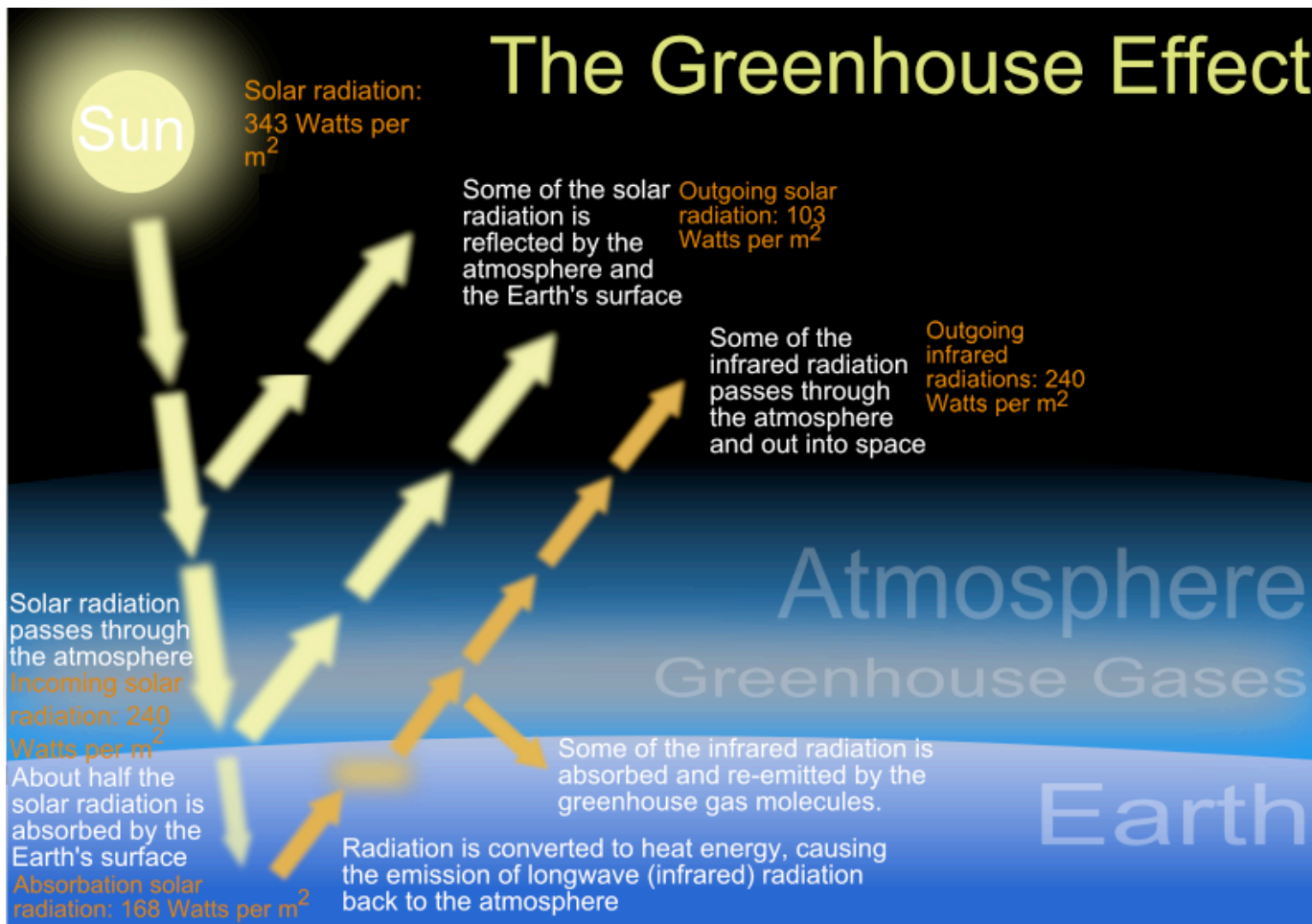
- $A, B > 0$
- $T \nearrow \Rightarrow \text{OLR} \nearrow$
- $\text{CO}_2 \leftrightarrow A ?$



Ray Pierrehumbert

*Principles of Planetary Climate*  
Cambridge, 2010  
(PPC)





- Top-of-the-atmosphere considerations
- Role of greenhouse gasses in reducing OLR

**Radiation**: characterized by direction of propagation and frequency

frequency  $\nu$  (Hz)

wavelength  $\lambda$  (m)

wavenumber  $n = \nu/c$   $m^{-1}$

$$\nu\lambda = c = 3 \times 10^8 \text{ m/s}$$

Wavelength (m)	Wavenumber ( $m^{-1}$ )	Frequency (Hz)		Median emission temperature (K)	Peak- $\nu$ temperature (K)	Peak- $\lambda$ temperature (K)
1000	0.001	$3 \times 10^5$	Radio	$4.1 \times 10^{-6}$	$5.1 \times 10^{-6}$	$2.9 \times 10^{-6}$
100	0.01	$3 \times 10^6$		$4.1 \times 10^{-5}$	$5.1 \times 10^{-5}$	$2.9 \times 10^{-5}$
10	0.1	$3 \times 10^7$		$4.1 \times 10^{-4}$	$5.1 \times 10^{-4}$	$2.9 \times 10^{-4}$
1	1	$3 \times 10^8$		$4.1 \times 10^{-3}$	$5.1 \times 10^{-3}$	$2.9 \times 10^{-3}$
0.1	10	$3 \times 10^9$	Microwave	0.041	0.051	0.029
0.01	100	$3 \times 10^{10}$		0.41	0.51	0.29
0.001	1000	$3 \times 10^{11}$	Infrared	4.1	5.1	2.9
$10^{-4}$	$10^4$	$3 \times 10^{12}$		41	51	29
$10^{-5}$	$10^5$	$3 \times 10^{13}$		410	510	290
$10^{-6}$	$10^6$	$3 \times 10^{14}$		4100	5100	2900
$10^{-7}$	$10^7$	$3 \times 10^{15}$	Visible	41 000	51 000	29 000
$10^{-8}$	$10^8$	$3 \times 10^{16}$	Ultra violet	$4.1 \times 10^5$	$5.1 \times 10^5$	$2.9 \times 10^5$
$10^{-9}$	$10^9$	$3 \times 10^{17}$	X-ray (soft)	$4.1 \times 10^6$	$5.1 \times 10^6$	$2.9 \times 10^6$
$10^{-10}$	$10^{10}$	$3 \times 10^{18}$		X-ray (hard)	$4.1 \times 10^7$	$5.1 \times 10^7$
$10^{-11}$	$10^{11}$	$3 \times 10^{19}$	Gamma ray	$4.1 \times 10^8$	$5.1 \times 10^8$	$2.9 \times 10^8$

Electromagnetic spectrum

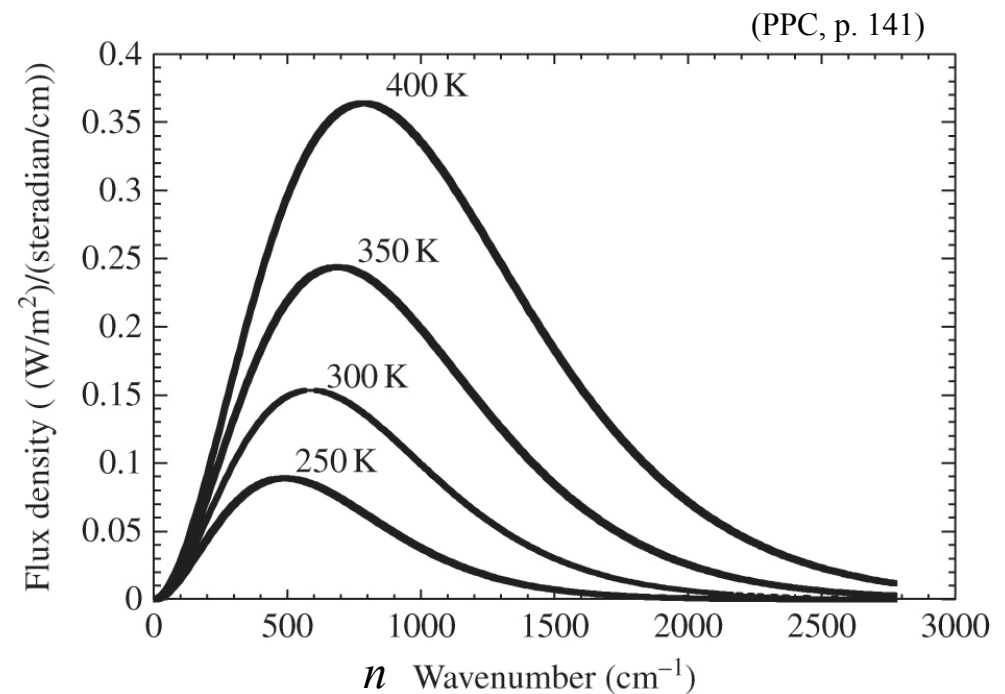
(PPC, p. 137)

## Blackbody radiation

- radiation reacts so strongly with matter that it achieves thermodynamic equilibrium at same temperature as the matter (“perfect absorbers and emitters”).

Planck’s Function: 
$$B(\nu, T) = \frac{2h\nu^3}{c^2} \frac{1}{e^{h\nu/kT} - 1}$$

$k$  = Boltzman thermodynamic constant  
 $h$  = Planck’s constant  
 $c$  = speed of light



- F.D.  $\nearrow$  with  $T$  at all  $n$
- F.D. small at large  $n$  for high  $T$
- Treat Earth as blackbody, even though core  $T \sim 6000$  K: sufficiently dense outer shell acts as blackbody

## Blackbody radiation

Stefan-Boltzman Law

Total power exiting from each unit area of the surface of a blackbody:

$$F = \int_0^{\infty} \pi B(\nu, T) d\nu = \sigma T^4,$$

$$\sigma = 2\pi^5 k^4 / (15c^2 h^3) \approx 5.67 \cdot 10^{-8} \text{ W/m}^2\text{K}^4$$

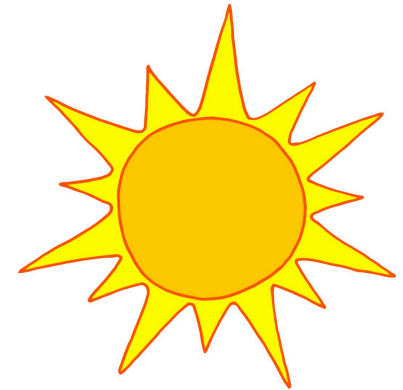
units of  $F$  = units of OLR =  $\text{W/m}^2$

## Radiation balance of planets: An idealized example

- only source of energy heating the planet is absorption of light from planet's host star.
- the planetary albedo is spatially uniform.
- the planet is spherical and has a distinct liquid or solid surface which radiates like a perfect blackbody.
- the temperature is uniform over the surface of the planet.
- the planet's atmosphere is perfectly transparent to the electromagnetic energy emitted by the surface



$a$  = planet's radius     $T_*$  = star's temperature  
 $r_*$  = star's radius     $r$  = distance to star



- Total flux impinging on planet  $\sigma T_*^4 r_*^2 / r^2 =$  solar constant  $L_*$
- Energy absorption:  $\pi a^2 L_* (1 - \alpha)$
- Energy loss:  $4\pi a^2 \sigma T^4$

Equilibrate:  $\sigma T^4 = \frac{L_*}{4} (1 - \alpha)$

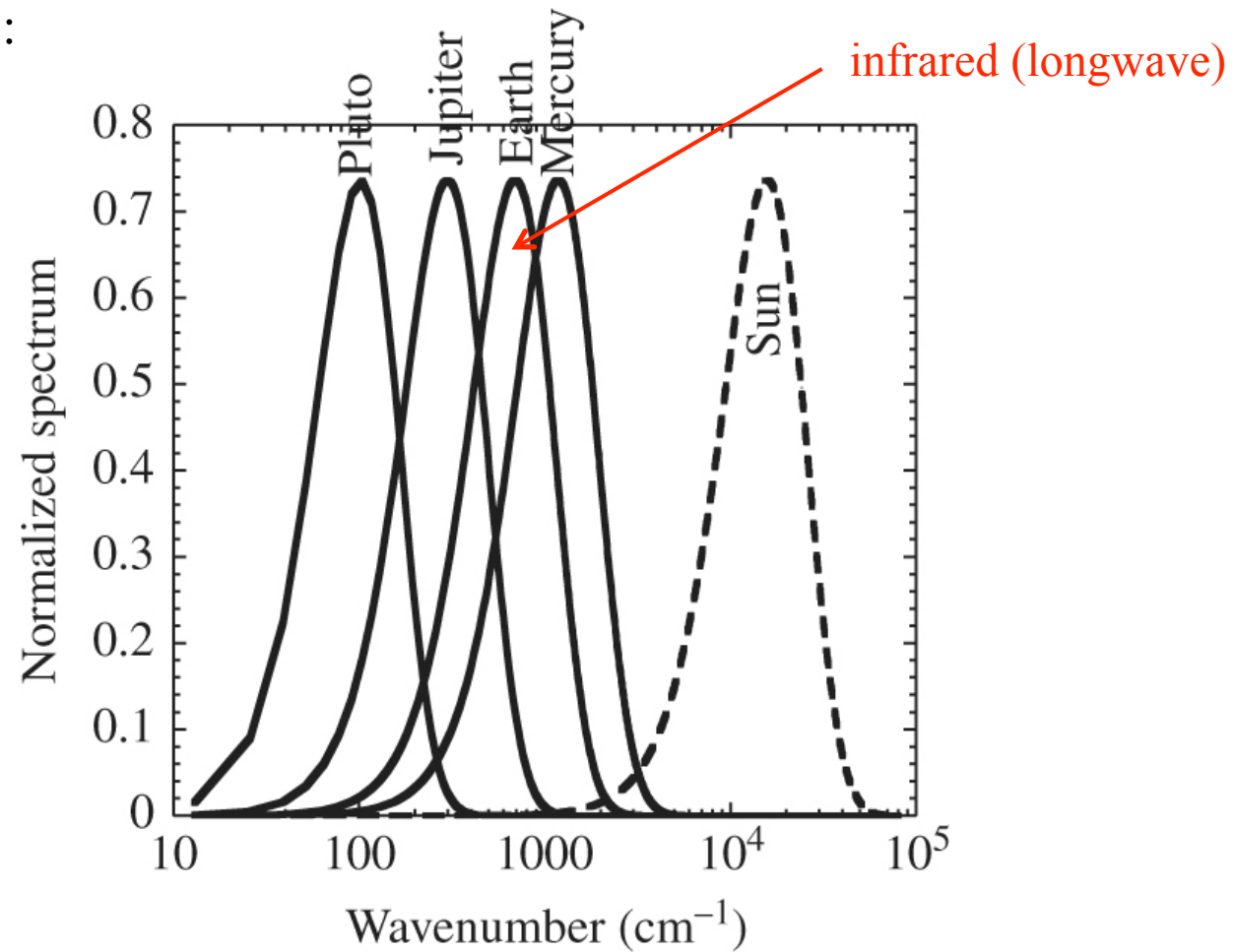
$$T = \frac{1}{\sqrt{2}} (1 - \alpha)^{1/4} \sqrt{\frac{r_*}{r}} T_*$$

Planet loses energy through emission at a lower wavenumber than that at which it receives energy from the star.



$$T = \frac{1}{\sqrt{2}}(1 - \alpha)^{1/4} \sqrt{\frac{r_*}{r}} T_*$$

Planck densities:



(PPC, p. 145)

## Greenhouse gas (G.G.): Basic ideas

- Earth radiates in the infrared
- If the atmosphere was transparent to infrared:  $OLR = \sigma T_s^4$
- Mix in G.G. with unit mass concentration  $q$

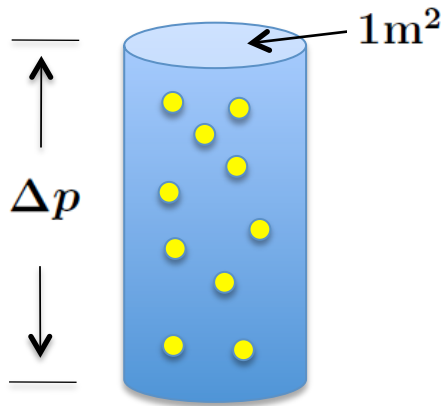
Assume G.G. (i) transparent to solar radiation

(ii) opaque to infrared at sufficiently high concentrations

$p_s$  = surface pressure

$T_s$  = surface temp

Assume  $T_s = T(p_s)$

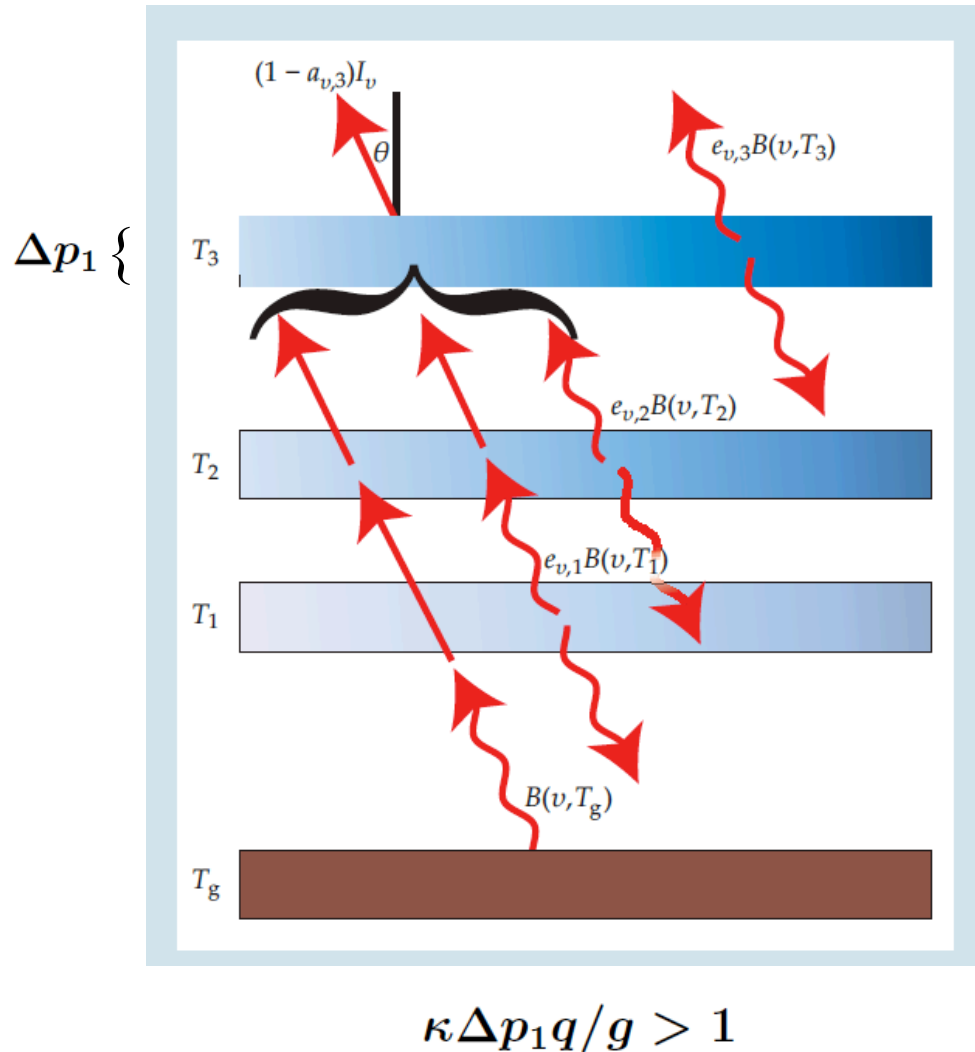


absorption coefficient  $\kappa = \kappa(\nu, p, T)$  m<sup>2</sup>/kg

$\kappa \Delta p q / g > 1 \Rightarrow$  column acts like blackbody

- $\kappa p_s q / g < 1 \Rightarrow$  atmosphere *optically thin*
- $\kappa p_s q / g \gg 1 \Rightarrow$  atmosphere *optically thick*

Greenhouse gas (G.G.): Optically thick case  $\kappa p_s q / g \gg 1$



OLR escapes only from top slab

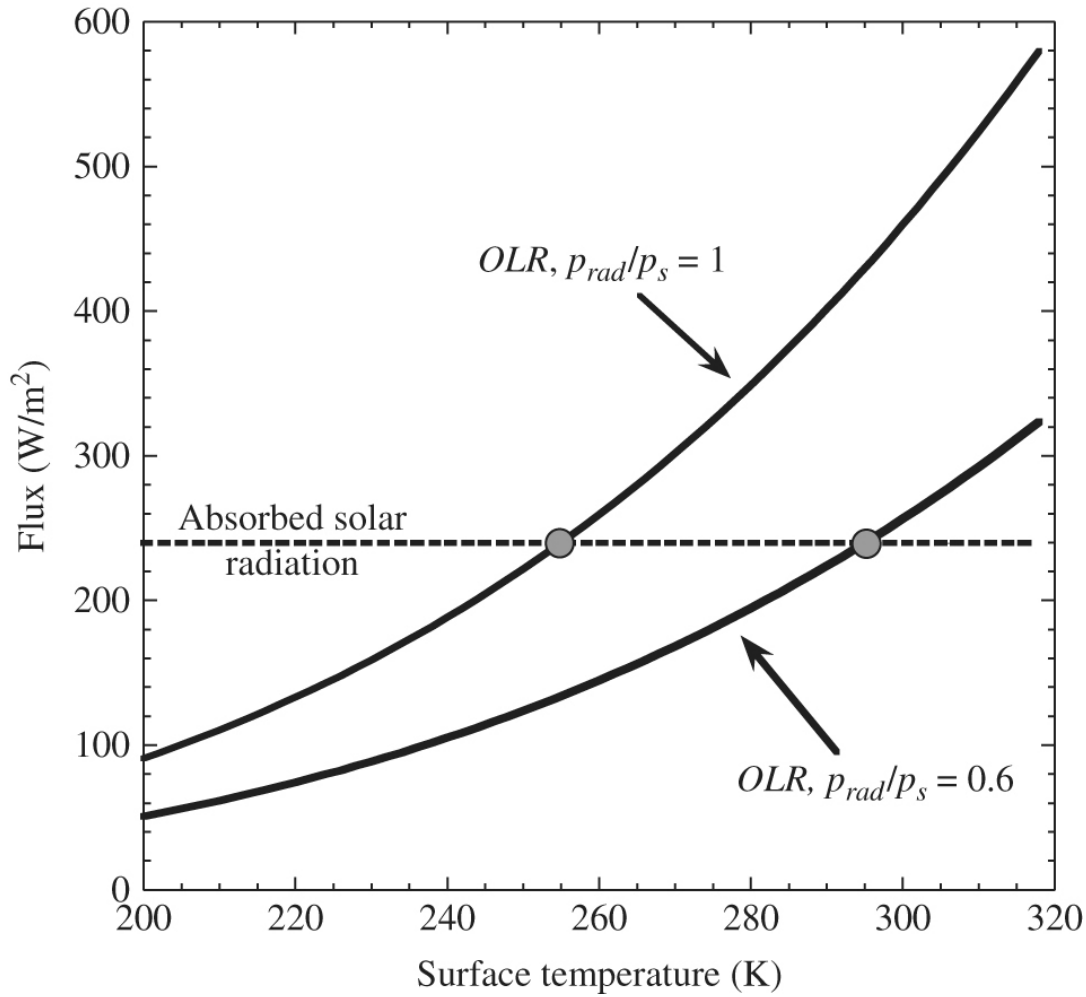
OLR determined by  $T_3$

$\Delta p_1 = p_{\text{rad}}$  characterizes the pressure at which OLR escapes

$$\text{OLR} \approx \sigma T(p_{\text{rad}})^4 < \sigma T_s^4$$

Figure from R. Pierrehumbert, Planetary radiation and planetary temperature, *Physics Today*, January 2011, 33-38.

## Greenhouse gas (G.G.): Basic ideas



(solar constant  $1370 \text{ W/m}^2$ , albedo 0.3)

(PPC, p. 147)

“In some sense, the whole subject of climate comes down to an ever-more sophisticated hierarchy of calculations of the curve  $OLR(T_s)$ .”

--R. Pierrehumbert, PPC, p. 146

## Greenhouse gas (G.G.): Basic ideas

$$I_s = \sigma T_0^4 [1 - m \tanh(19 T_0^6 \times 10^{-16})]$$

*atmospheric transmission factor*

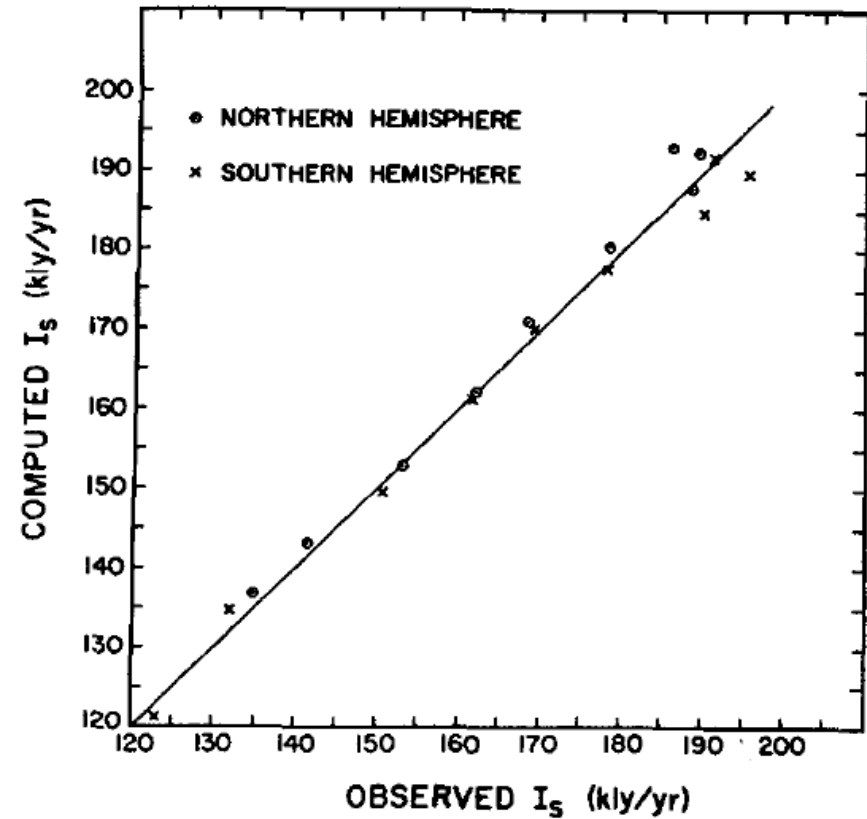
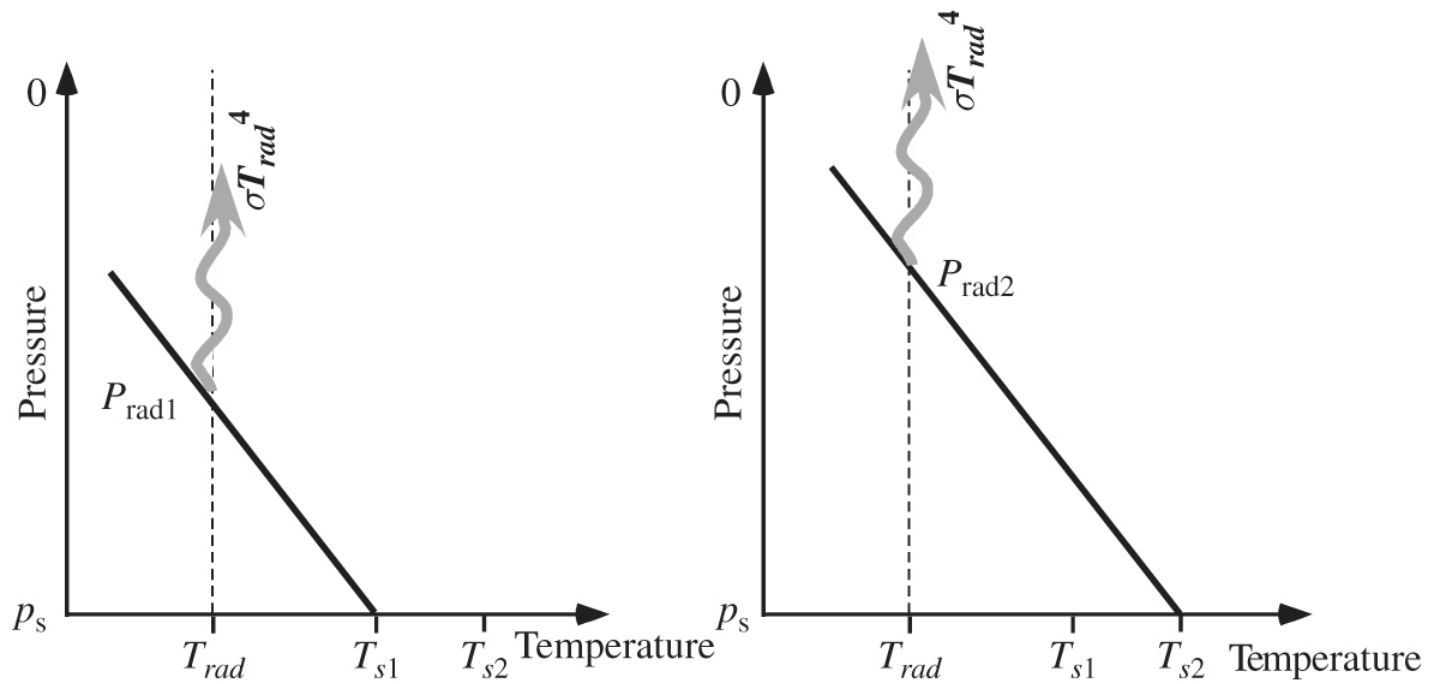


FIG. 1. A plot of the computed and observed values of the annual infrared emission to space in  $10^\circ$  latitude belts. The computed values were obtained from Eq. (6). The observed values are those given by Sellers (1965), modified slightly by recent satellite measurements.

W. Sellers, A global climate model based on the energy balance of the earth-atmosphere system, *J. Appl. Meteorology* **8** (1969), 392-400.

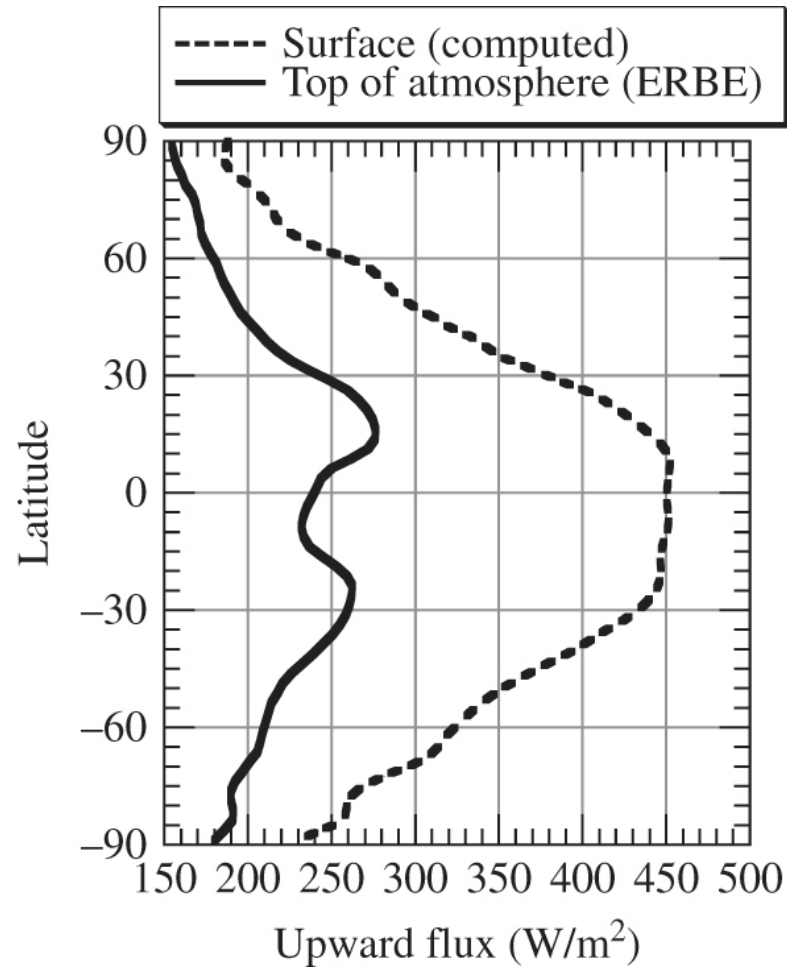
## Greenhouse gas (G.G.): Basic ideas

- $\sigma T_{\text{rad}}^4 = \frac{1}{4}(1 - \alpha)L_*$
- $p_{\text{rad}} \searrow$  as G.G.  $\nearrow$



(Figure from PPC, p. 147)

## Greenhouse gas (G.G.): Basic ideas



(Figure from PPC, p. 149)

The Earth's observed zonal-mean OLR for January, 1986 (solid curve).

$\sigma T_s^4$  (dashed curve).

## Greenhouse gas (G.G.): Basic ideas

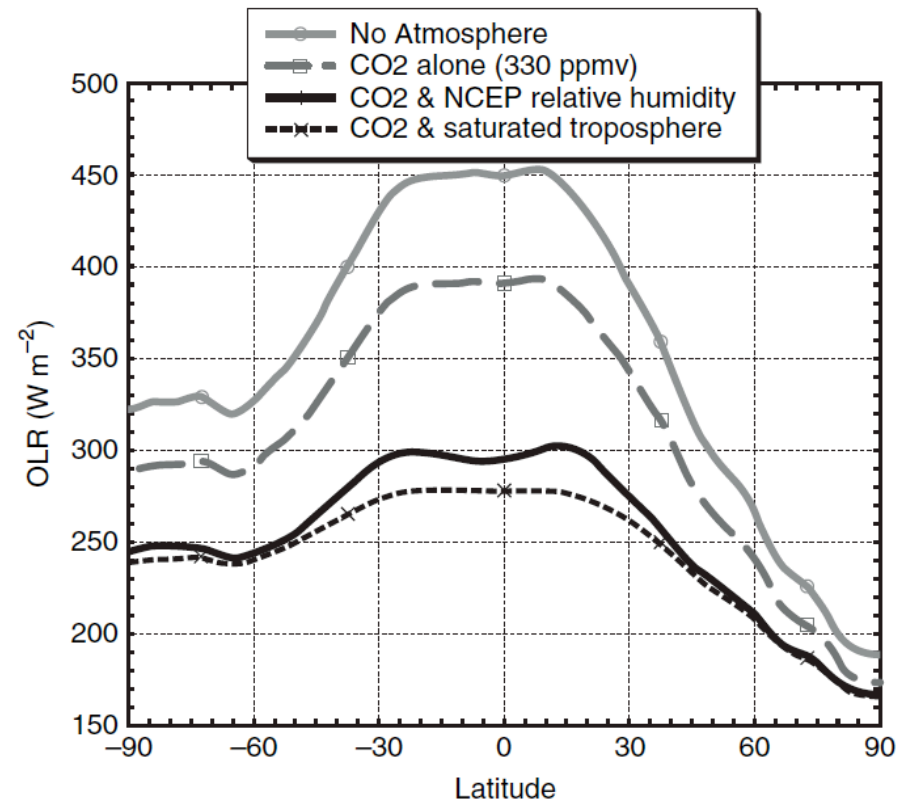
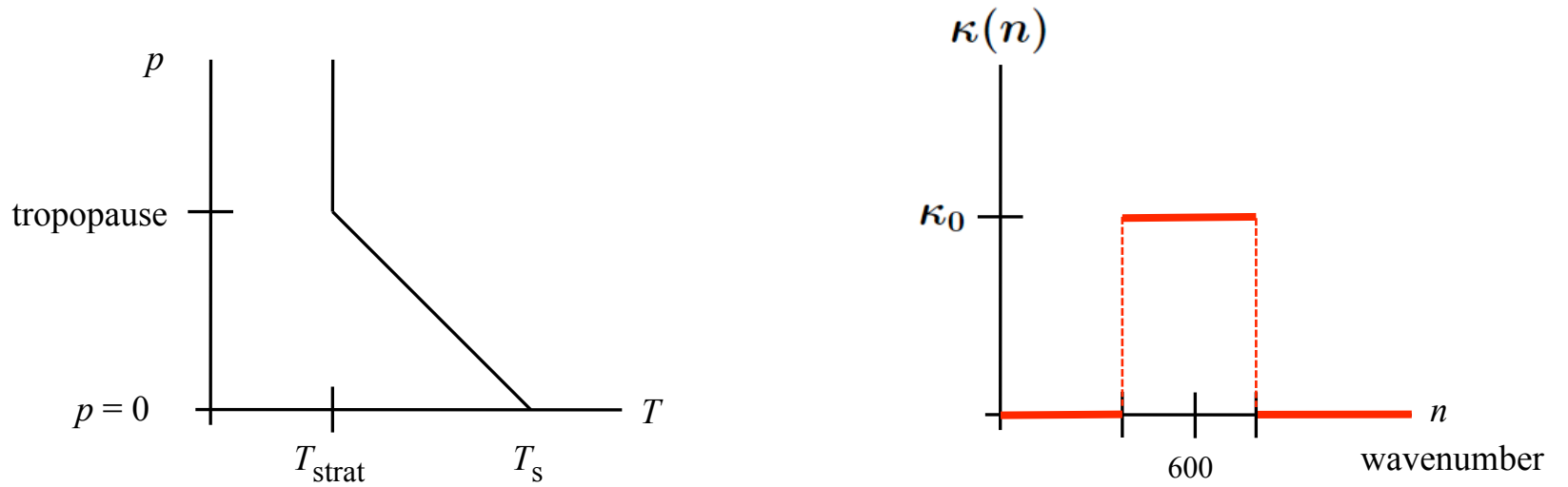


FIGURE 6.1. The effect of atmospheric composition on OLR, with atmospheric and surface temperature held fixed at January climatological values, 1960–1980.

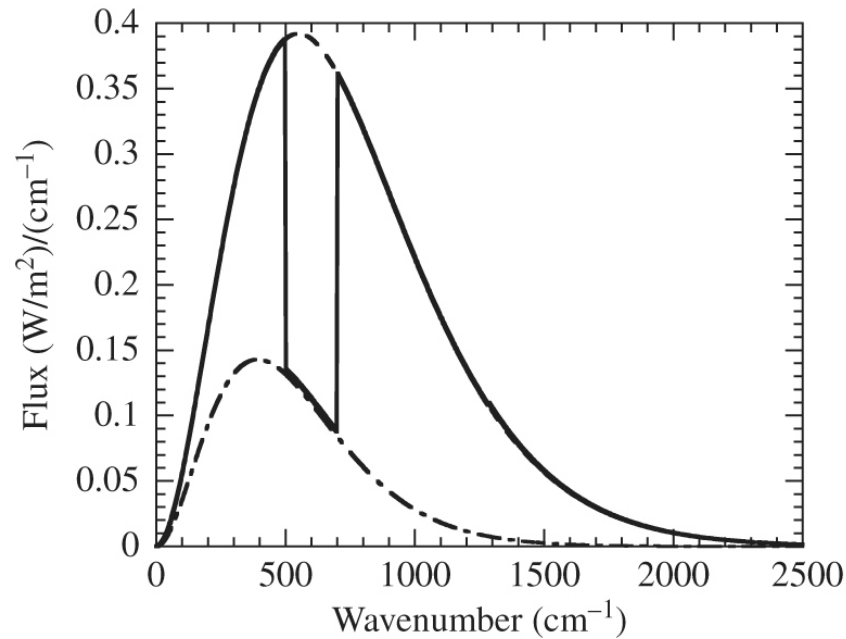
R. Pierrehumbert et al, On the relative humidity of the atmosphere, *The Global Circulation of the Atmosphere*, T. Schneider and A Sobel, eds. Princeton University Press, 2007, p. 143.



OLR spectrum: Toy example with one (fictitious) G.G. *oobleck*

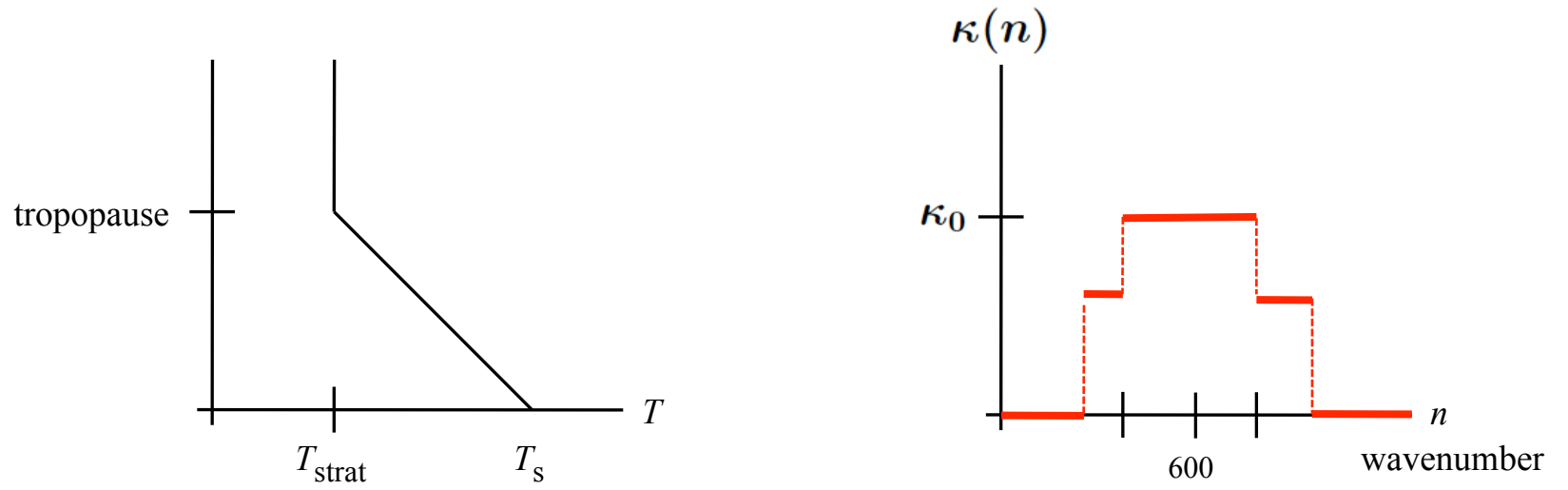


$(T_s = 280 \text{ K}, T_{\text{strat}} = 200 \text{ K})$

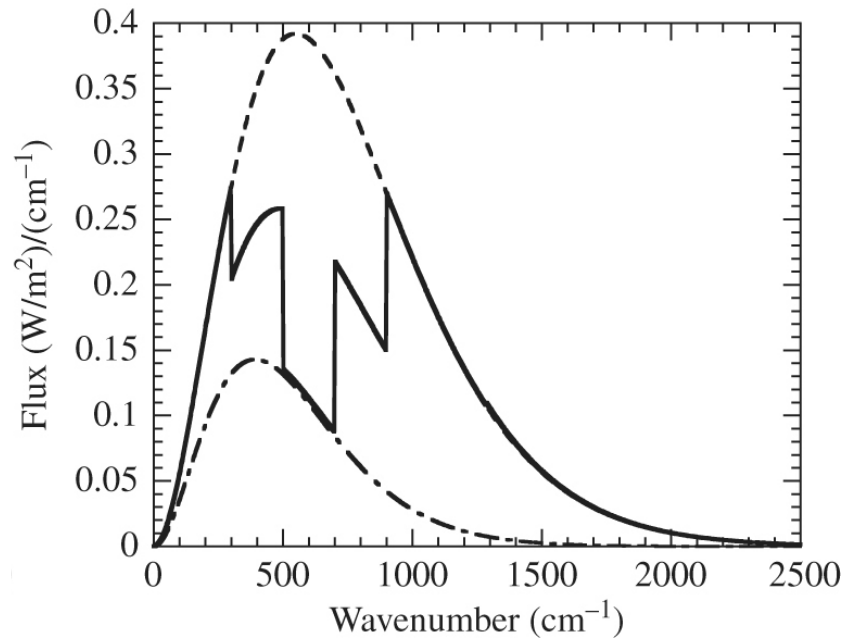


(Figure from PPC, p. 218)

OLR spectrum: Toy example with one (fictitious) G.G. *em*

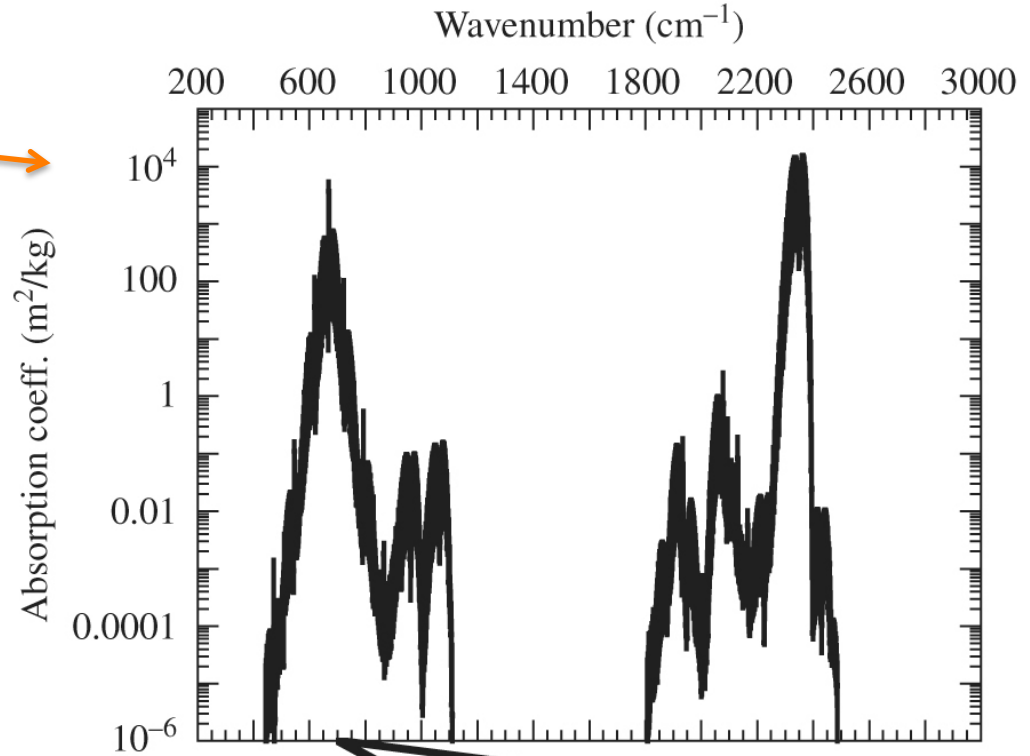


$(T_s = 280 \text{ K}, T_{\text{strat}} = 200 \text{ K})$

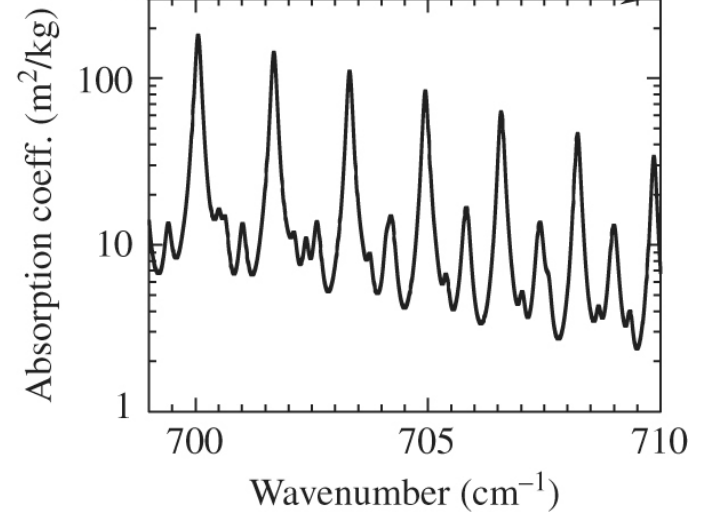
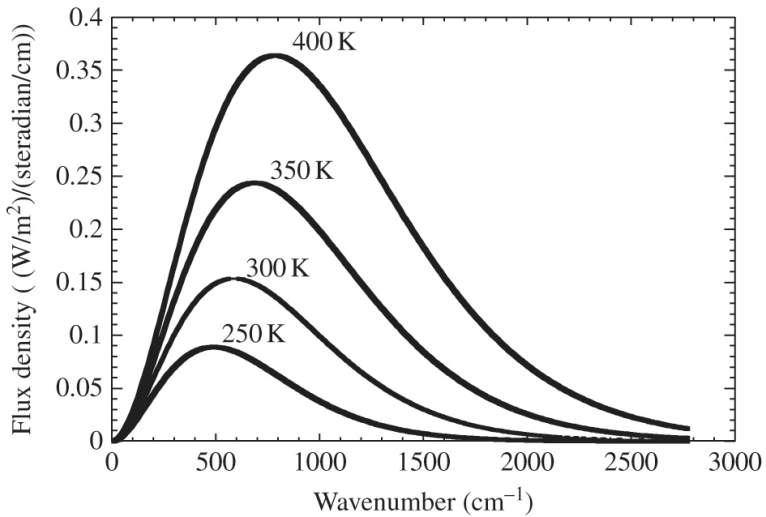


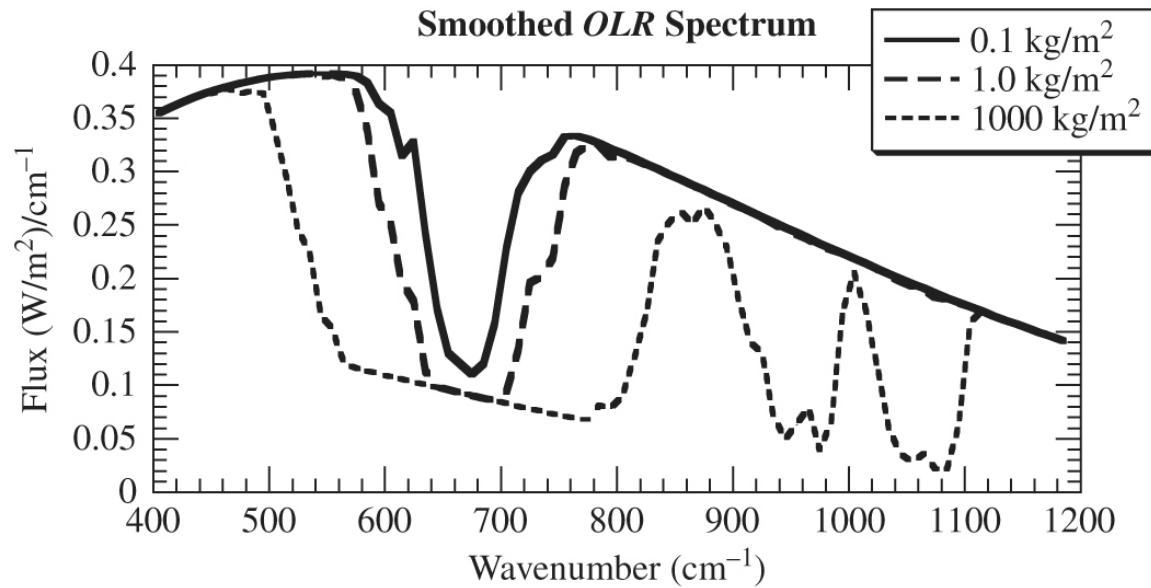
(Figure from PPC, p. 218)

Absorption coefficient  
for pure CO<sub>2</sub> at  
 $T = 293 \text{ K}$ ,  $p = 1000 \text{ mb}$



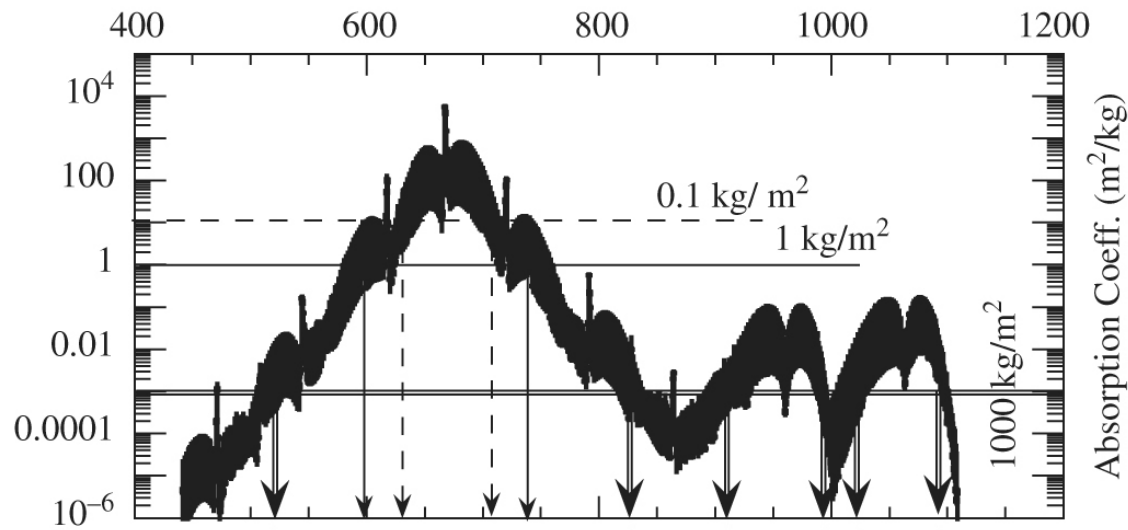
Infrared emission peak for Earth  
centered on  $\sim n = 670 \text{ cm}^{-1}$



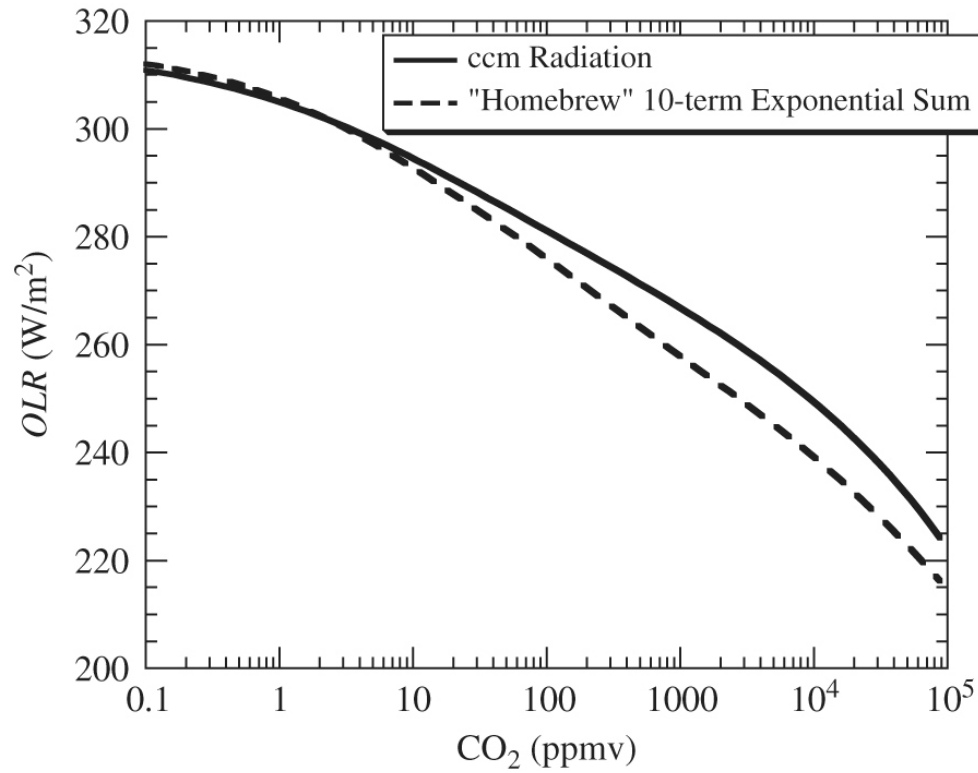


“The rate at which absorption decays with distance determines the rate at which the OLR decreases as the greenhouse gas concentration is made larger.”

--R. Pierrehumbert, PPC, p. 219



Absorption coefficient for CO<sub>2</sub> at 1 bar and 300K (bottom). Corresponding OLR for the three concentrations (top).

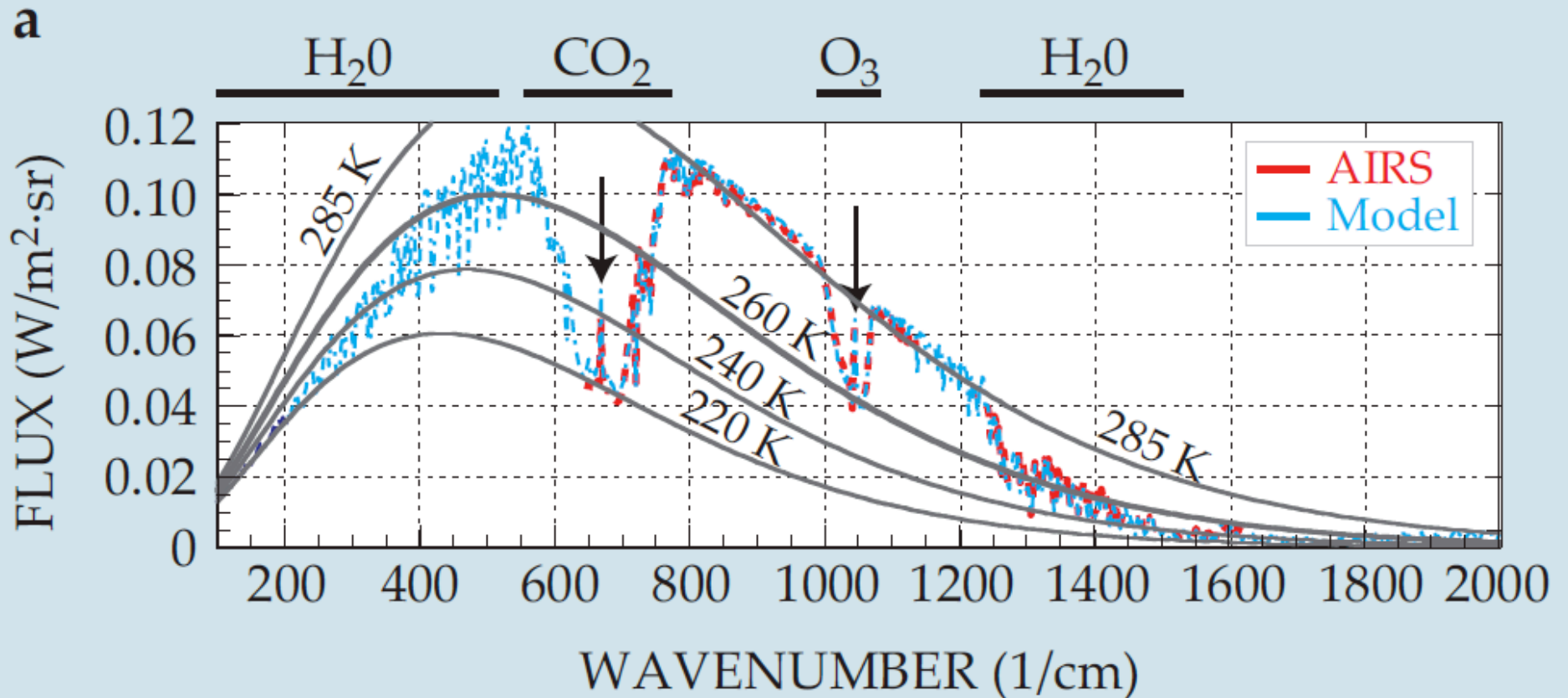
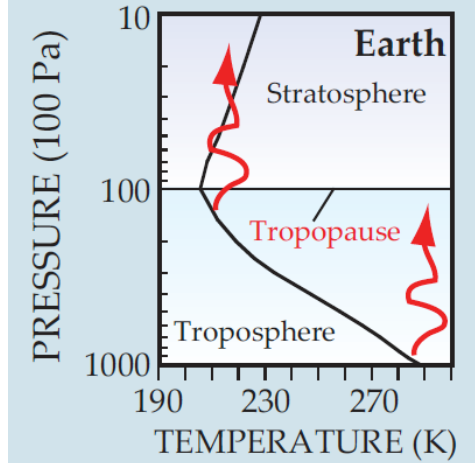


The  $OLR$  vs  $CO_2$  concentration for  $CO_2$  in a dry air atmosphere.  $T_s = 273K$  held fixed. This curve gives the amount of absorbed solar radiation needed to maintain the surface temperature at freezing.

$OLR$  goes down approximately in proportion to the logarithm of the  $CO_2$  concentration.

R. Pierrehumbert, Infrared radiation and planetary temperature, *Physics Today* (January 2011).

The left panel compares a computed global-mean, annual-mean emission spectrum for Earth (blue) with observations from the satellite-borne AIRS instrument (red); both are superimposed over a series of Planck distributions.



$$R \frac{\partial T}{\partial t} = Q_s(y)(1 - \alpha(y)) - (A + BT) - C \left( T - \int_0^1 T dy \right)$$

$T$   $5^\circ \times 5^\circ$

$OLR$   $15^\circ \times 15^\circ$

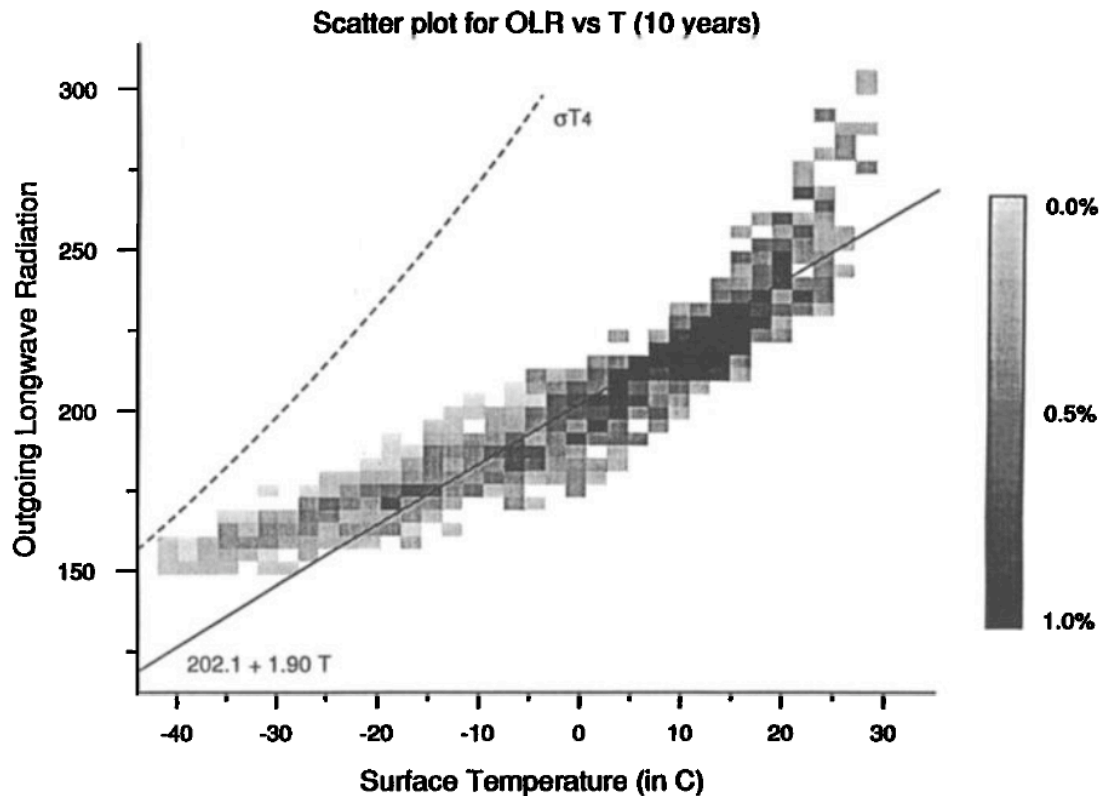


Fig. 2. Scatter plot of OLR versus surface temperature from  $30^\circ\text{N}$  to  $90^\circ\text{N}$  from the 10-year data set. The scale on the right indicates the percent of total. Note that there is a cosine of latitude weighting to account for the differing grid point areas.

C. Graves et al, New parameterizations and sensitivities for simple climate models, *J. Geophysical Research* **98** (1993), 5025-5036.

Insolation =  $OLR$

( $T$  global average temperature)

$$Q = \sigma T^4$$

$$Q = 240 \Rightarrow T \approx 255K$$

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$$Q = \sigma T^4 - \bar{G} = \sigma T^4 - 150 \Rightarrow T \approx 288K$$

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Let time vary:

$$R \frac{dT}{dt} = Q - OLR = Q - (\sigma T(t)^4 - G(t)) \quad G(t) = \bar{G} + \beta \ln \left( \frac{C}{C_0} \right)$$

$C = C(t)$  = atmospheric concentration of  $CO_2$  (ppm)

*After* A. Hogg, Glacial cycles and carbon dioxide: A conceptual model, *Geophysical Res. Letters* **35** (2008)



$$R \frac{dT}{dt} = Q - OLR = Q - (\sigma T(t)^4 - G(t)) \quad G(t) = \bar{G} + \beta \ln \left( \frac{C}{C_0} \right)$$

$C = C(t) =$  atmospheric concentration of CO<sub>2</sub> (ppm)

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Linear approximation:  $T_0 = 273$  ,  $T$  °C,  $T/T_0 \ll 1$

$$T^4 = (T_0 + T)^4 = T_0^4(1 + T/T_0)^4 \approx T_0^4(1 + 4T/T_0)$$

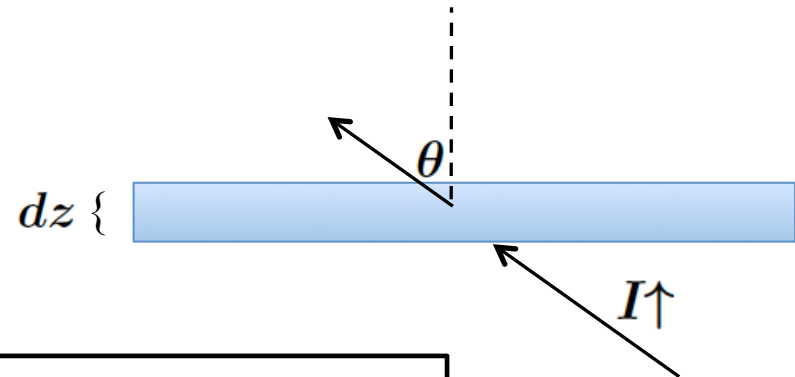
Kelvin! 

$$OLR = \sigma T^4 - G \approx \sigma T_0^4 + 4\sigma T_0^3 T - \left( \bar{G} + \beta \ln \left( \frac{C}{C_0} \right) \right) = BT + A,$$

$$A = A(t) = \sigma T_0^4 - \left( \bar{G} + \beta \ln \left( \frac{C(t)}{C_0} \right) \right)$$

After A. Hogg, Glacial cycles and carbon dioxide: A conceptual model, *Geophysical Res. Letters* **35** (2008)

Basic equations for radiative flux\*



$$dI_{\uparrow} = -\kappa \rho_r I_{\uparrow} \sec \theta dz + \kappa \rho_r I_b \sec \theta dz$$

*Schwarzschild equation for radiative transfer without scattering*

- $z$  height above surface
- $\kappa$  absorption coefficient
- $\rho_r$  density of radiating gas
- $I_b$  blackbody radiation flux from Planck

Energy absorbed from the incoming beam by the absorbing gas.  
Energy emitted by the layer of gas.

- assume: given  $z$ ,  $T$  and  $p$  are known
- consider  $\kappa = \kappa(\nu, z)$
- consider  $I_b = I_b(\nu, z)$

\*G. Plass, Infrared radiation in the atmosphere, *Amer. J. Physics* **24** (1956), 303-321.

## Basic equations for radiative flux

$$dI_{\uparrow} = -\kappa \rho_r I_{\uparrow} \sec \theta dz + \kappa \rho_r I_b \sec \theta dz$$

new independent variable: mass of radiating gas per unit area

$$u = \int_z^{\infty} \rho_r dz = \int_z^{\infty} c \rho dz, \quad du = -\rho_r dz$$

$\rho$  total density of atmosphere at height  $z$

$c = \rho_r / \rho$  fractional concentration of radiating gas

$u = 0$  at t.o.a

$$dI_{\uparrow} = \kappa I_{\uparrow} \sec \theta du - \kappa I_b \sec \theta du$$

- $\kappa = \kappa(u)$
- $I_b = I_b(u)$

## Basic equations for radiative flux

$$dI_{\uparrow} = \kappa I_{\uparrow} \sec \theta du - \kappa I_b \sec \theta du$$

$$u = \int_z^{\infty} \rho_r dz = \int_z^{\infty} c \rho dz$$

Transmission function between “heights”  $u_0 < u_1$  :

$$\tau(u_0, u_1) = \exp \left( - \sec \theta \int_{u_0}^{u_1} \kappa du \right)$$

- $\kappa \approx 0 \Rightarrow \tau \approx 1$
- $\kappa \gg 1 \Rightarrow \tau \approx 0$

Solution: 
$$I_{\uparrow}(u) = I_{\uparrow}(u_1) \tau(u, u_1) + \sec \theta \int_u^{u_1} \kappa(v) I_b(v) \tau(u, v) dv$$

upward flux at the lower boundary layer  $u_1$  times the transmission between  $u$  and  $u_1$

blackbody flux emitted by each layer of gas between  $u$  and  $u_1$  times the transmission between  $u$  and the height of the emitting layer

## Basic equations for radiative flux

$$u = \int_z^\infty \rho_r dz = \int_z^\infty c \rho dz$$

$$\tau(u_0, u_1) = \exp\left(-\sec \theta \int_{u_0}^{u_1} \kappa du\right)$$

Solution:  $I\uparrow(u) = I\uparrow(u_1)\tau(u, u_1) + \sec \theta \int_u^{u_1} \kappa(v) I_b(v) \tau(u, v) dv$

Remove explicit dependence on absorption coefficient: Integrate by parts!

$$I\uparrow(u) = I\uparrow(u_1)\tau(u, u_1) + I_b(u) - I_b(u_1)\tau(u, u_1) + \int_u^{u_1} \tau(u, v) \frac{dI_b(v)}{dv} dv$$

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Downward flux:  $dI\downarrow = -\kappa I\downarrow \sec \theta du + \kappa I_b \sec \theta du$

$$I\downarrow(u) = I\downarrow(u_0)\tau(u_0, u) + I_b(u) - I_b(u_0)\tau(u_0, u) - \int_{u_0}^u \tau(v, u) \frac{dI_b(v)}{dv} dv$$

## Basic equations for radiative flux

$$I\uparrow(u) = I\uparrow(u_1)\tau(u, u_1) + I_b(u) - I_b(u_1)\tau(u, u_1) + \int_u^{u_1} \tau(u, v) \frac{dI_b(v)}{dv} dv$$

$$I\downarrow(u) = I\downarrow(u_0)\tau(u_0, u) + I_b(u) - I_b(u_0)\tau(u_0, u) - \int_{u_0}^u \tau(v, u) \frac{dI_b(v)}{dv} dv$$

$u_1$  value of  $u$  at lower boundary (usually the surface)

$u_0$  value of  $u$  at upper boundary (usually t.o.a, so  $u_0 = 0$ )

Approximation:  $I\downarrow(0) = 0$      $I\uparrow(u_1) = I_b(u_1)$



solar infrared flux that reaches  
Earth very small relative to OLR



Earth's surface radiates as  
(almost) perfect blackbody in the  
infrared

## Basic equations for radiative flux

$$I\uparrow(u) = I_b(u) + \int_u^{u_1} \tau(u, v) \frac{dI_b(v)}{dv} dv$$

$$I\downarrow(u) = I_b(u) - I_b(0)\tau(0, u) - \int_0^u \tau(v, u) \frac{dI_b(v)}{dv} dv$$

Infrared radiation does not come from a single level—a bit is contributed from each level (each having its own  $T$ ).

A bit of this is absorbed at each intervening level of the atmosphere.

Radiation is emitted in all directions.

Rate of emission and absorption strongly depends on frequency.

Several greenhouse gasses.

Currently, Earth receives about  $1 \text{ W/m}^2$  more from solar absorption than it emits to space as infrared (due to rapid rise of greenhouse gases).

Earth's temperature has not yet risen enough to restore the energy balance: takes time to first warm up the oceans and melt the ice. (PPC, p. 152)